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## Development of Optical Trace Gas monitoring Technology for NASA Human Space Flight

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**OUTLINE**

- Motivation: Gas sensing applications
- Quartz Enhanced Laser-PAS
- Selected Applications of QE-PAS
  - NH<sub>3</sub> Detection with 1.5 μm RT cw DFB Diode Laser
  - H<sub>2</sub>CO Detection with 3.5 μm CW DFB Interband Cascade Laser
- Tunable mid-IR laser sources
  - External Cavity QCL design
  - High resolution spectroscopy of NO using EC-QCL
- Summary and future directions

HABITATION 2008  
 5-8 February, 2008  
 Orlando, FL

## Motivation: Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
  - Industrial Plants
  - Combustion Sources and Processes (eg. early fire sensing)
  - Automobile and Aircraft Emissions
- **Rural Emission Measurements**
  - Agriculture and Animal Facilities
- **Environmental Monitoring**
  - Atmospheric Chemistry (e.g. ecosystems and airborne)
  - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
  - Chemical, Pharmaceutical, Food & Semiconductor Industry
- **Spacecraft and Planetary Surface Monitoring**
  - **Crew Health Maintenance & Human Life Support Program**
- **Medical Diagnostics** (eg. breath analysis)
- **Biohazard and Toxic Chemical Detection**
- **Homeland Security**
- **Fundamental Science and Photochemistry**

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## Fundamentals of Laser Absorption Spectroscopy

**B Beer-Lambert's Law of Linear Absorption**

$$I(\nu) = I_0 e^{-\alpha(\nu) P L}$$

$\alpha(\nu)$  - absorption coefficient [ $\text{cm}^{-1} \text{atm}^{-1}$ ];  $L$  - path length [cm]  
 $\nu$  - frequency [ $\text{cm}^{-1}$ ];  $P$  - partial pressure [atm]

$\alpha(\nu) = C \cdot S(T) \cdot g(\nu - \nu_0)$

$C$  - total number of molecules of absorbing gas/atm  $\text{cm}^3$  [molecule  $\text{cm}^{-3}$  atm $^{-1}$ ]  
 $S$  - molecular line intensity [cm $^2$  molecule $^{-1}$ ]  
 $g(\nu - \nu_0)$  - normalized spectral lineshape function [cm], (Gaussian, Lorentzian, Voigt)

**Optimum Molecular Absorbing Transition**

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

**Long Optical Pathlengths**

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & waveguides)

**Spectroscopic Detection Schemes**

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- **Photoacoustic Spectroscopy**

## From conventional PAS to QEPAS

Laser beam, power  $P$

Modulated ( $P$  or  $\lambda$ ) at  $f$  or  $f/2$

$$S \sim \frac{Q \alpha P}{f V}$$

$$NNEA = \frac{\alpha_{\min} P}{\sqrt{\Delta f}} \left[ \frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}} \right]$$

$Q \gg 1000$   
 Cell is **OPTIONAL!**  
 Effective volume  
 Bioelectric crystal microphone  
 Resonant at  $f$   
 quality factor  $Q$

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## Diode laser based Quartz-Enhanced Photoacoustic Spectrometer

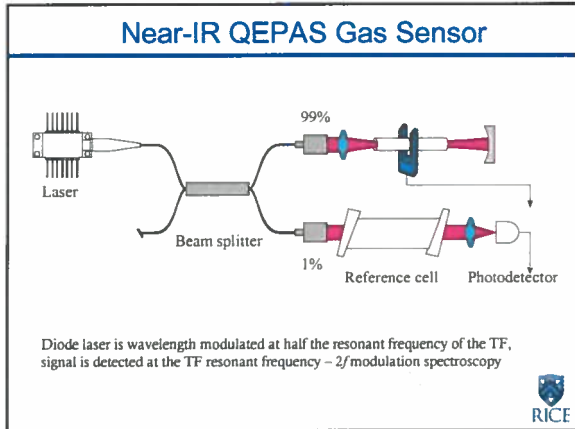
A. Kosterev et al. Appl. Phys. B 80, 133 (2005)

## Comparative Size of Absorbance Detection Modules (ADM)

Optical multipass cell (100 m):  
 $l \sim 70 \text{ cm}$ ,  $V \sim 3000 \text{ cm}^3$

Resonant photoacoustic cell (1000 Hz):  
 $l \sim 60 \text{ cm}$ ,  $V \sim 50 \text{ cm}^3$

QEPAS ADM:  
 $l \sim 0.5 \text{ cm}$ ,  $V \sim 0.05 \text{ cm}^3$



### NASA Target Gas Opportunity Matrix

Molecule	Detection Limit (ppb)	QEPAS detectable?	
		1.3-1.7 $\mu\text{m}$	2-5 $\mu\text{m}$
Formaldehyde	10	No	X
Acetaldehyde	20	Experiments required	
Ammonia	100	X	X
Carbon monoxide	1000	Probably not	X
Hydrogen cyanide	100	X	X
Carbon dioxide	<2%	X	X
Nitrogen dioxide	100	Probably not	X
HF	100	Experiments required	
Acrolein (2-Propenal)	5	Unlikely	
Water vapor	10-90%	X	X

X - Demonstrated  
 X - Highly expected based on the existing technology level  
 X - Expected with the technology advance

### QEPAS Performance for 10 Trace Gas Species (Dec'05)

Molecule (Host)	Frequency, $\text{cm}^{-1}$	Pressure, Torr	NNEA, $\text{cm}^{-1}\text{W}/\text{Hz}^2$	Power, mW	NEC ( $\tau=1\text{s}$ ), ppmv
H <sub>2</sub> O (N <sub>2</sub> )**	7181.17	60	$2.1 \times 10^{-3}$	5.8	0.18
HCN (air: 50% hum) **	6529.11	60	$< 2.6 \times 10^{-3}$	50	0.1
C <sub>2</sub> H <sub>2</sub> (N <sub>2</sub> )**	6529.17	75	$-2.5 \times 10^{-3}$	-40	0.06
NH <sub>3</sub> (N <sub>2</sub> )*	6528.76	60	$5.4 \times 10^{-3}$	38	0.50
CO <sub>2</sub> (exhaled air)	6514.25	90	$1.0 \times 10^{-3}$	5.2	890
CO <sub>2</sub> (N <sub>2</sub> )***	4990.00	300	$1.5 \times 10^{-3}$	4.6	130
CH <sub>4</sub> O (N <sub>2</sub> )*	2832.48	100	$1.1 \times 10^{-3}$	4.6	0.28
CO (N <sub>2</sub> )	2196.66	50	$5.3 \times 10^{-3}$	13	0.5
CO (propylene)	2196.66	50	$7.4 \times 10^{-3}$	6.5	0.14
N <sub>2</sub> O (air+5%SF <sub>6</sub> )	2195.63	50	$1.5 \times 10^{-3}$	19	0.007

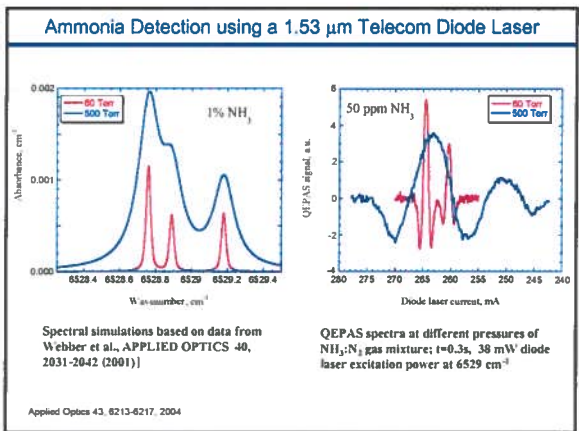
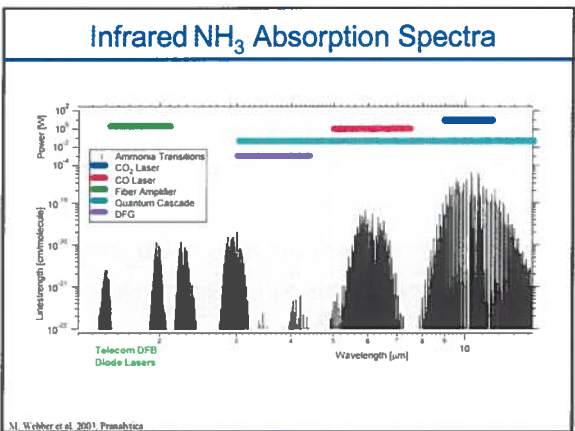
\* - Improved microresonator  
 \*\* - Improved microresonator and double optical pass through QTF  
 \*\*\* - Without microresonator

NNEA - normalized noise equivalent absorption coefficient.  
 NEC - noise equivalent concentration for available laser power and  $\tau=1\text{s}$  time constant.

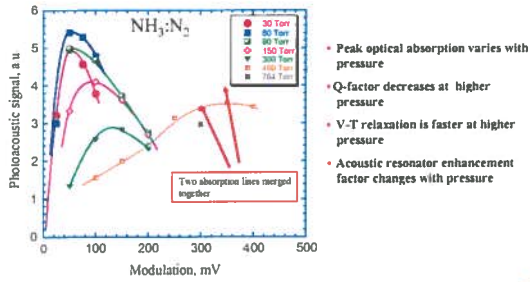
For comparison: conventional PAS  $2.2 \times 10^{-6} \text{ cm}^{-1}\text{W}/\sqrt{\text{Hz}}$  (1,800 Hz) for NH<sub>3</sub>\*

\* M. E. Webber, M. Paulkanky and C. K. N. Patel, Appl. Opt. 42, 2119-2128 (2003)

- ### Motivation for NH<sub>3</sub> Detection
- Spacecraft related gas monitoring
  - Monitoring NH<sub>3</sub> concentrations in the exhaust stream of NO<sub>x</sub> removal systems based on selective catalytic reduction (SCR) techniques
  - Semiconductor process monitoring & control
  - Monitoring of industrial refrigeration facilities
  - Pollutant gas monitoring
  - Atmospheric chemistry
  - Medical diagnostics (kidney & liver dysfunctions)



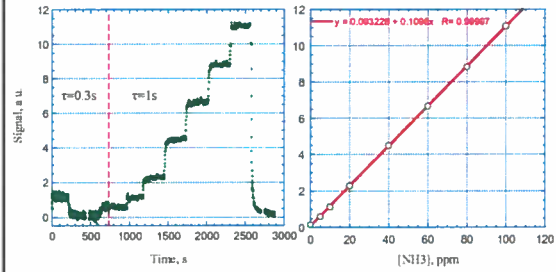
### Pressure Dependence of QEPAS Sensitivity



- Peak optical absorption varies with pressure
- Q-factor decreases at higher pressure
- V-T relaxation is faster at higher pressure
- Acoustic resonator enhancement factor changes with pressure



### Calibration and Linearity of QEPAS based NH<sub>3</sub> Sensor

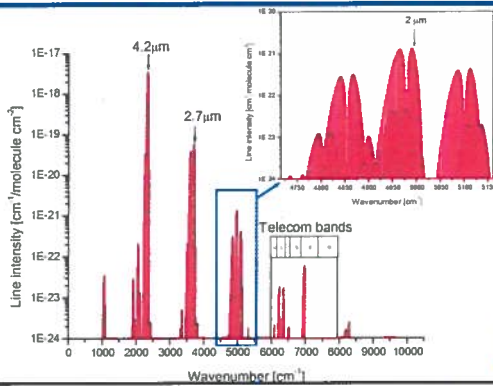


Noise-equivalent (1s) concentration (NEC) for  $\tau=1s$  time constant is 0.65 ppmv for 38 mW excitation power

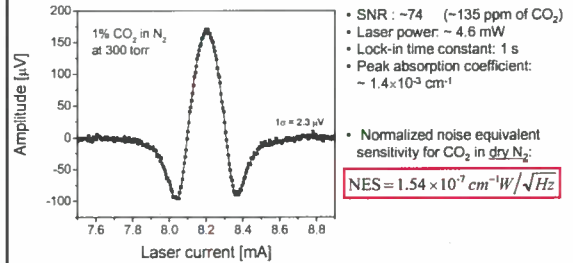
90 last points of each step averaged  
(Traditional PAS\* -  $2.2 \times 10^{-6} \text{ cm}^{-1} \text{ W/Hz}$ )  
\*Weaver et al., APPLIED OPTICS-42, 2118, 2003

To Date Noise-equivalent Absorption (NEA) Coefficient  $k=5.4 \times 10^{-7} \text{ cm}^{-1} \text{ W/Hz}^{1/2}$

### CO<sub>2</sub> absorption spectrum



### CO<sub>2</sub> detection limit



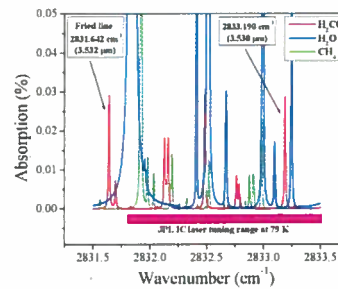
Normalized noise equivalent sensitivity for CO<sub>2</sub> in dry N<sub>2</sub>:  
 $NES = 1.54 \times 10^{-7} \text{ cm}^{-1} \text{ W} / \sqrt{\text{Hz}}$



### Motivation for Precision Monitoring of H<sub>2</sub>CO

- Pollutant due to incomplete fuel combustion processes
- Potential trace contaminant in industrial manufactured products
- Precursor to atmospheric O<sub>3</sub> production
- Medically important gas

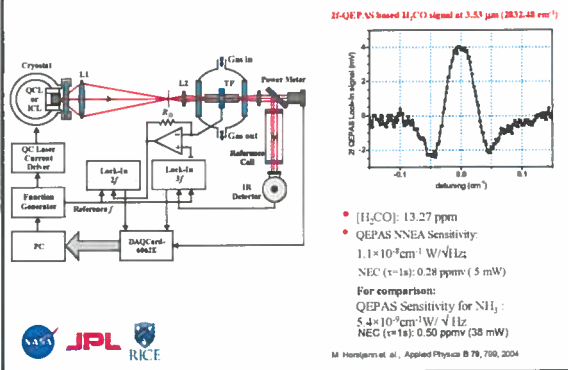
### HITRAN Based Simulation of a H<sub>2</sub>CO-H<sub>2</sub>O-CH<sub>4</sub> Spectrum in Tuning Range of a 3.53 micrometers Laser



- H<sub>2</sub>CO : 10 ppb
- H<sub>2</sub>O : 3%
- CH<sub>4</sub> : 2 ppm
- Optical path: 100 m
- Total pressure: 30 Torr



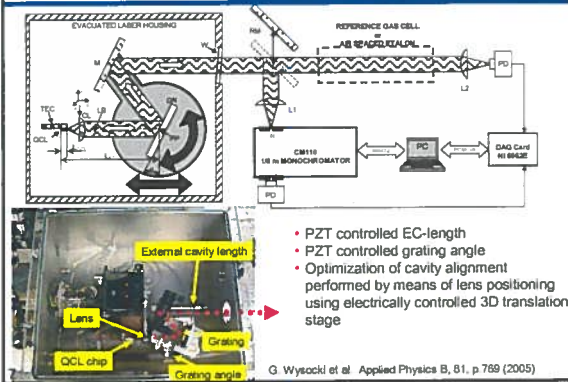
## QCL based Quartz-Enhanced Photoacoustic Sensor



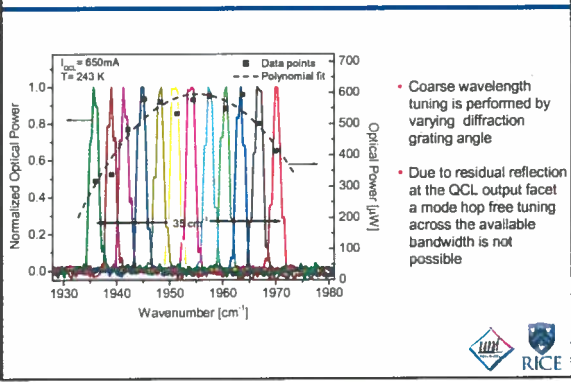
## Merits of QE Laser-PAS based Trace Gas Detection

- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- Immunity to ambient and flow acoustic noise, laser noise and etalon effects, which allows applications that involve harsh operating environments
- Required sample volume is very small. The volume is ultimately limited by the gap size between the TF prongs, which is < 1 mm<sup>3</sup> for the presently used QTF.
- No spectrally selective elements are required
- Applicable over a wide range of pressures, including atmospheric pressure
- Sensitive to phase shift introduced by vibrational to translational (V-T) relaxation processes and hence the potential of concentration measurements of spectrally interfering species
- Ultra-compact, rugged and low cost compared to LAS that requires a multipass absorption cell and infrared detector(s)
- Potential for optically multiplexed concentration measurements

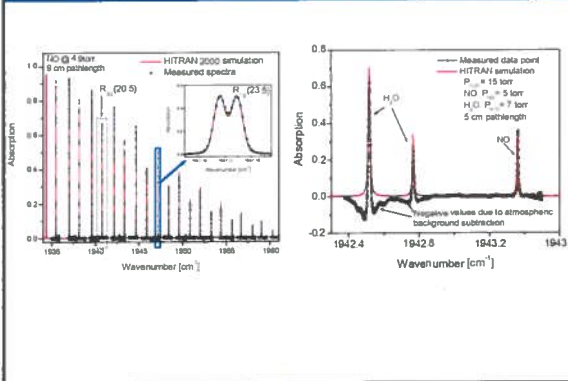
## External Cavity QCL Based Spectrometer



## Wide Wavelength Tuning with Diffraction Grating



## Mid-IR NO Absorption Spectra Acquired with a Tunable TEC QCL



## Summary and Future Directions

- **QEPAS based Trace Gas Sensors**
  - Ultra compact (~ 0.2 mm<sup>3</sup>), robust & low cost sensors based on QEPAS
  - QEPAS is immune to ambient noise. The measured noise level coincides with the thermal noise of the QTF
  - Best to date demonstrated QEPAS sensitivity is 2.1 × 10<sup>-4</sup> cm<sup>-1</sup>W/√Hz for H<sub>2</sub>O:N<sub>2</sub>
  - QEPAS exhibits a low 1/f noise level, allowing data averaging for more than 3 hours
  - Detected 14 trace gases to date: NH<sub>3</sub>, CH<sub>4</sub>, N<sub>2</sub>O, CO<sub>2</sub>, CO, NO, H<sub>2</sub>O, COS, HCN, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>5</sub>OH, SO<sub>2</sub>, H<sub>2</sub>CO and several isotopic species of C, O, N & H
  - Sensitive to phase shift introduced by V-T relaxation dynamics
  - Potential for spectrally interfering species concentration measurements
- **Applications in Trace Gas Detection**
  - Environmental & Spacecraft Monitoring (NH<sub>3</sub>, CO, CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>, N<sub>2</sub>O, CO<sub>2</sub> and H<sub>2</sub>CO)
  - Medical Diagnostics (NO, CO, COS, CO<sub>2</sub>, NH<sub>3</sub>, C<sub>2</sub>H<sub>4</sub>)
  - Industrial process control and chemical analysis (NO, NH<sub>3</sub>, H<sub>2</sub>O)
- **Future Directions and Collaborations**
  - QEPAS based applications using novel thermoelectrically cooled cw and broadly wavelength tunable quantum and interband cascade lasers
  - Investigate QTFs with lower resonant frequencies
  - Investigate amplitude modulation QEPAS potential and limitations
  - New target gases, in particular VOCs and HCs
  - Development of optically multiplexed gas sensor networks based on QEPAS