

RICE **Advanced Infrared Semiconductor Laser based Chemical Sensing Technologies**

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<http://ece.rice.edu/lasersci/>

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- Motivation: Wide Range of Chemical Sensing
- Fundamentals of Laser Absorption Spectroscopy
- New laser sources and sensing technologies
- Selected Applications of Trace Gas Detection
 - Detection of nitric oxide and ethanol
 - Quartz Enhanced L-PAS (Freon 125, acetone & ammonia)
- Future Directions and Conclusions

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Wide Range of Trace Gas Sensing Applications

- Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- Environmental Monitoring**
 - Atmospheric Chemistry
 - Volcanic Emissions
- Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Nuclear Safeguards, Pharmaceutical, Metals Processing, Food & Beverage Industries
- Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- Applications in Health and Life Sciences**
- Technologies for Law Enforcement and National Security**
- Fundamental Science and Photochemistry**

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Fundamentals of Laser Absorption Spectroscopy

Beer-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_s L}$
 $\alpha(\nu)$ - absorption coefficient [$\text{cm}^{-1} \text{atm}^{-1}$]; L - path length (cm)
 ν - frequency [cm^{-1}]; P_s - partial pressure (atm)

Key Requirements: Sensitivity, specificity, rapid data acquisition and multi-species detection

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & hollow waveguides)

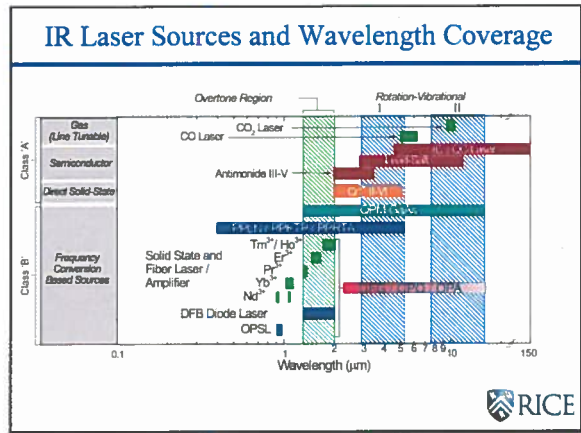
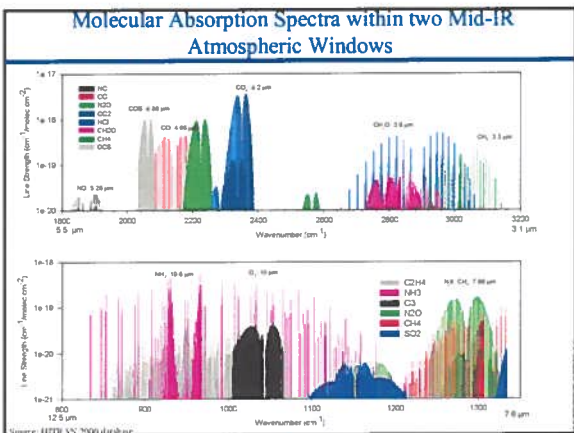
Spectroscopic Detection Schemes

- Wavelength & Frequency Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

C - total number of molecules of absorbing gas/ atm cm^2 [molecule $\text{cm}^{-1} \text{atm}^{-1}$]
 S - molecular line intensity [cm molecule $^{-1}$]
 $g(\nu - \nu_0)$ - normalized spectral lineshape function [cm], (Gaussian, Lorentzian, Voigt)

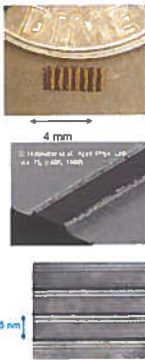
Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to ppt)	Wavelength, Power
Selectivity (Spectral Resolution)	Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Tunable Wavelength
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	No Consumables
Field deployable	Compact & Robust



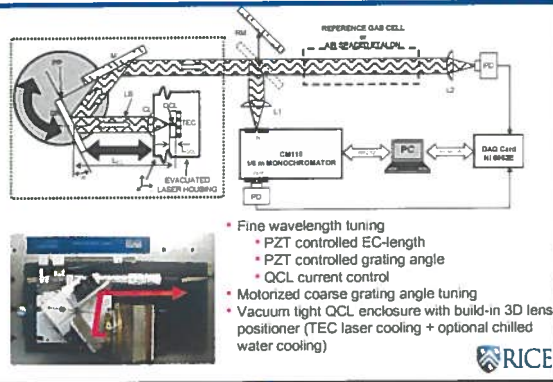
Key Characteristics of mid-IR QCLs and ICL Sources

- Band - structure engineered devices** (emission wavelength is determined by layer thickness - MBE or MOCVD); mid-infrared QCLs operate from 2 to 24 μm
- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength
- Spectral tuning range in the mid-IR** (4-24 μm for QCLs and 3-5 μm for ICLs)
 - 1.5 cm^{-1} using injection current control
 - 10-20 cm^{-1} using temperature control
 - > 265 cm^{-1} using an external grating element and with heterogeneous cascade active region design
- Narrow spectral linewidth** (cw: 0.1 - 3 MHz & < 10 kHz with frequency stabilization @ 0004 cm^{-1}), pulsed - 300 MHz (chirp from heating)
- High pulsed and cw powers at TEC/RT temperatures**
 - Pulsed peak powers of 1.6 W, high temperature operation \sim 425K
 - Average power levels: 1-600 mW (current wall plug $\eta \sim 1\%$)
 - ~ 50 mW, TEC CW DFB @ 5 and 10 μm Alpes, Princeton, Adtech Optics, Maxon Technologies, Hamamatsu, Daylight
 - ~ 300 mW @ 8.3 μm (Agilent Technologies & Harvard)
 - > 600 mW (CW FP) @ RT & a wall plug efficiency of \sim 3%, > 150 mW (CW DFB) at 298 K (Northwestern)

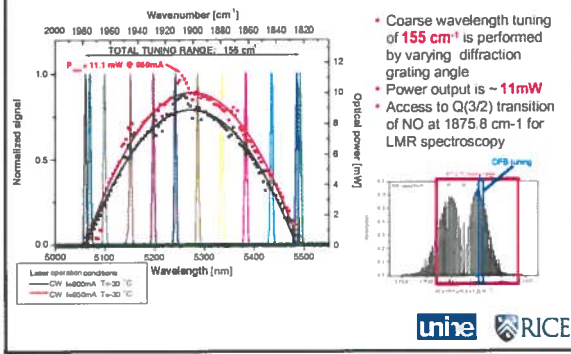


Widely Tunable, CW, TEC Quantum Cascade Lasers

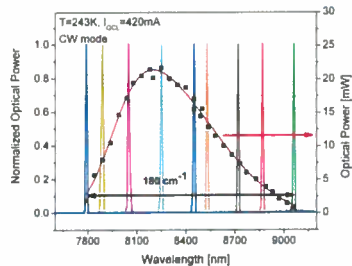
Tunable external cavity QCL based spectrometer



Wide Wavelength Tuning of a 5.3 μm EC-QCL



Performance of 8.4 μm EC-QCL Spectroscopic Source



Tunability **180 cm^{-1}** @ 8.4 μm (1100 to 1280 cm^{-1})

AR coating: $R_{AR} = 2 \times 10^{-4}$ P_{EC-opt} up to **50 mW (cw)**
 (I_{loc} = 680 mA \rightarrow P = 44 mW)



Quartz Enhanced Photoacoustic Spectroscopy

From conventional PAS to QEPAS

Laser beam, power P

Modulated (P or λ) at f or $f/2$

$S \sim \frac{Q \alpha P}{f V}$

$NNEA = \frac{\alpha_{min} P}{\sqrt{\Delta f}} \left[\frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}} \right]$

Cell is OPTIONAL!

SWAP RESONATING ELEMENT!!!

Effective volume

$Q \gg 1000$

Piezoelectric crystal

Resonant at f quality factor Q

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Quartz Tuning Fork (TF) as a Resonant Microphone

- Resonant frequency $f=32.8$ kHz
- Intrinsically high Q factor: $Q_{\text{resonator}} = 125\,000$, $Q_{\text{air}} = 10\,000$ at ambient conditions.
- Piezoelectric: requires no transducer
- Miniature size
- Mass produced for clocks – low cost

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Absorption Detection Module for QEPAS based Gas Sensor

$\text{Ø}0.41$ mm

10 mm

3.6 mm

Lens

Excitation laser beam

Quartz tuning fork electrodes

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Comparative Size of Absorption Detection Modules (ADM)

- Resonant frequency $f = 32.8$ kHz
- Intrinsically high Q factor: $Q_{\text{resonator}} = 125\,000$, $Q_{\text{air}} = 10\,000$ for ambient conditions
- Piezoelectric: requires no transducer
- Miniature size
- Mass produced for watches & clocks – low cost

Optical multipass cell (100 m):
 $l=70$ cm, $V=3000$ cm³

Resonant photoacoustic cell (1000 Hz):
 $l=60$ cm, $V=50$ cm³

QEPAS spectrophone:
 $l=1$ cm, $V=0.05$ cm³

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Alignment-free QEPAS Absorption Detection Module

Quartz Tuning Fork

Acoustic Micro Resonator

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Merits of QE Laser-PAS based Trace Gas Detection

- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- Immune to ambient and flow acoustic noise, laser noise and etalon effects
- Significant reduction of sample volume (< 1 mm³)
- Applicable over a wide range of pressures
- Temperature, pressure and humidity insensitive
- Rugged and low cost (compared to other optical sensor architectures)

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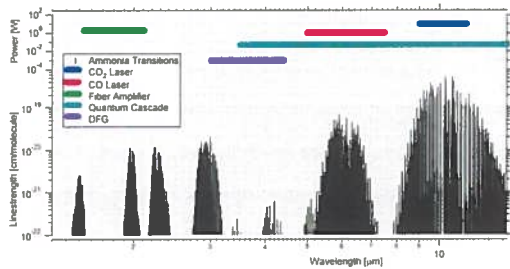
Trace Gas Sensing Examples

Motivation for NH₃ Detection

- Monitoring of gas separation processes
- Spacecraft related gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Pollutant gas monitoring
- Atmospheric chemistry
- Medical diagnostics (kidney & liver dysfunctions)

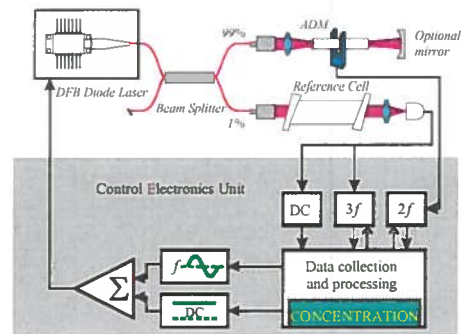


Infrared NH₃ Absorption Spectra



M. Webber et al. 2003. PNAS

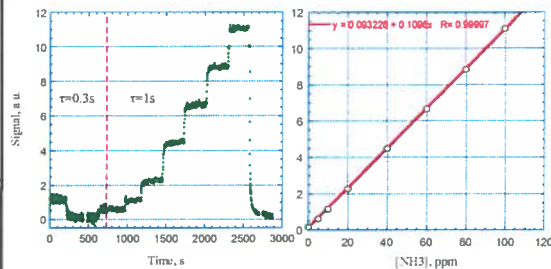
QEPAS based Gas Sensor Architecture



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Calibration and Linearity of QEPAS based NH₃ Sensor



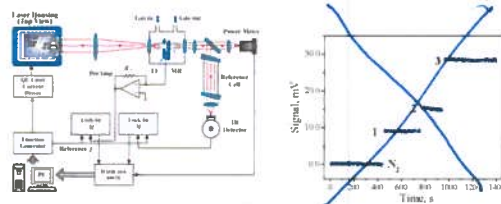
Noise-equivalent (1s) concentration (NEC), for $\tau=1s$ time constant is 0.65 ppm for 38 mW excitation power

90 last points of each step averaged

Noise-equivalent absorption (NEA) coefficient $k = 2.4 \times 10^{-6} \text{ cm}^{-1} \text{ W/Hz}^{1/2}$



CW DFB QCL based QEPAS Ammonia Sensor operating at 1046.4 cm⁻¹



Motivation for Nitric Oxide Detection

- Atmospheric Chemistry
- Environmental pollutant gas monitoring
 - NO_x monitoring from automobile exhaust and power plant emissions
 - Precursor of smog and acid rain
- Industrial process control
 - Formation of oxynitride gates in CMOS Devices
- NO in medicine and biology
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, COPD, acute lung rejection



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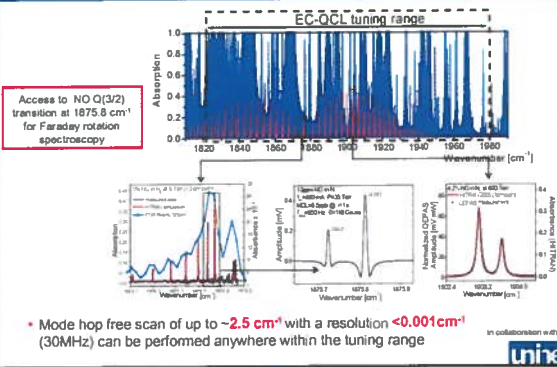
Biomarkers Present in Exhaled Human Breath

More than 400 different molecules in breath; many with well defined biochemical pathways

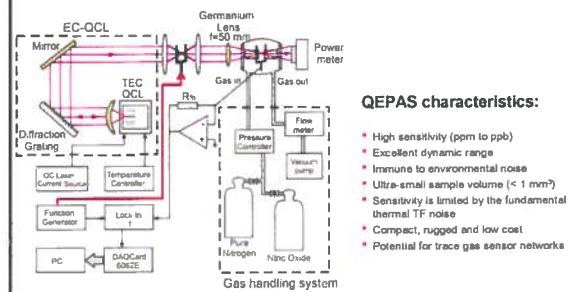
BROADBAND ABSORBERS	Compound	Concentration	Physiological basis/Pathology Indication
→	Acetaldehyde	ppb	Ethanol metabolism
→	Ammonia	ppm	Decarboxylation of acetate; diabetes, protein metabolism, liver and renal disease
→	Acetone	ppb	Product of respiration; Helicobacter pylori
→	Carbon dioxide	%	Gut bacteria, schizophrenia
→	Carbon disulfide	ppb	Production catalyzed by ferredoxinase
→	Carbon monoxide	ppm	Gut bacteria, liver disease
→	Carbonyl sulfide	ppb	Lipid peroxidation and oxidative stress
→	Ethane	ppb	Gut bacteria
→	Ethanol	ppb	Lipid peroxidation, oxidative stress, cancer
→	Ethylene	ppb	Lipid peroxidation/metabolism
→	Hydrocarbons	ppb	Gut bacteria
→	Hydrogen	ppm	Cholesterol biosynthesis
→	Isoprene	ppb	Gut bacteria
→	Methane	ppm	Methionine metabolism
→	Methanethiol	ppb	Metabolism of fruit
→	Methanol	ppb	Protein metabolism
→	Methylamine	ppb	Protein metabolism
→	Nitric oxide	ppb	Production catalyzed by nitric oxide synthase
→	Oxygen	%	Required for normal respiration
→	Pentane	ppb	Lipid peroxidation, oxidative stress
→	Water	%	Product of respiration

Terence Rabby, Johns Hopkins University

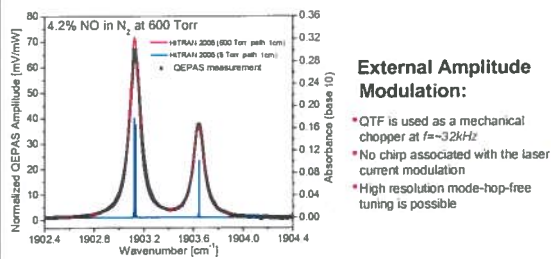
High resolution spectroscopy with a 5.3μm EC-QCL



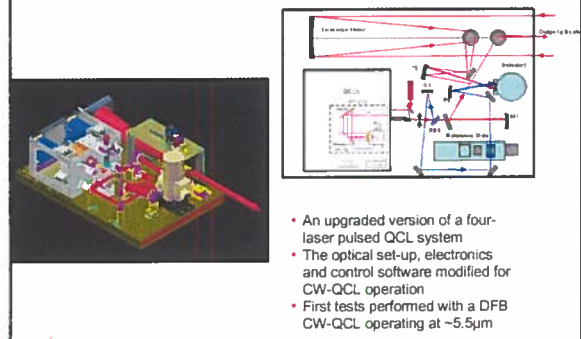
QCL based Quartz-Enhanced Photoacoustic Gas Sensor

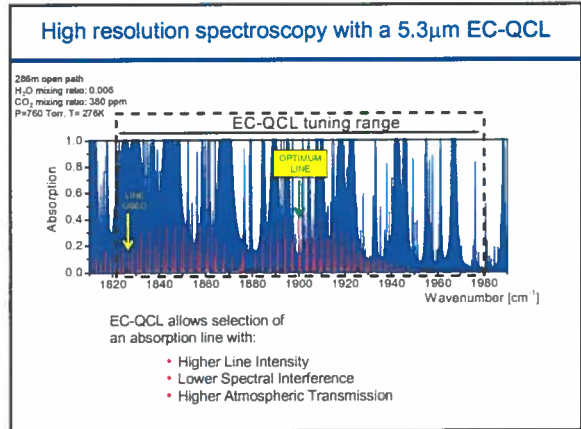
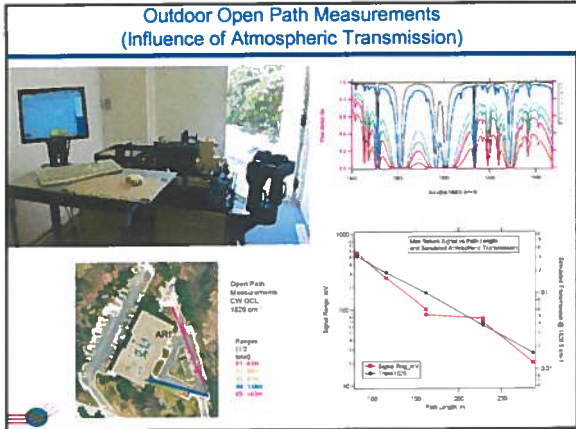


High resolution EC-QCL based QEPAS



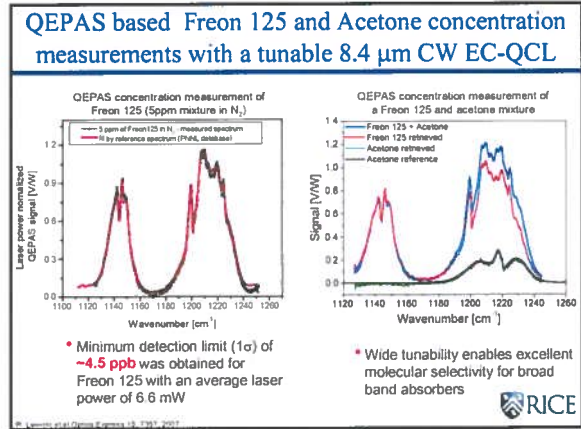
Design of an EC-QCL Based Remote Sensing System





red sees laser

- ### Monitoring of broadband absorbers
- Freon 125 (C₂F₅)
 - Refrigerant (leak detection)
 - Safe simulant for toxic chemicals, e.g. chemical warfare agents
 - Acetone (CH₃COCH₃)
 - Recognized biomarker for diabetes
- RICE



QEPAS Performance for 12 Trace Gas Species (Feb '08)

Molecule (list)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ⁻¹ W/Hz ²	Power, mW	NEC (σ=1s), ppb·m
H ₂ O (N ₂) ^{***}	7306.75	60	1.5×10 ¹⁰	9.3	0.09
HCN (air=96% RB) ^{**}	6339.11	60	4.3×10 ¹⁰	50	0.16
C ₂ H ₂ (N ₂) ^{**}	6323.88	720	4.1×10 ¹⁰	37	0.03
NH ₃ (N ₂) ^{**}	6328.76	573	3.1×10 ¹⁰	60	0.06
C ₂ H ₄ (N ₂) ^{**}	6177.07	715	5.4×10 ¹⁰	15	1.7
CH ₄ (N ₂) ^{**}	6057.09	930	2.9×10 ¹⁰	13.7	2.1
CO ₂ (breath=100% RB)	6061.23	150	8.2×10 ¹⁰	45	4.0
H ₂ S (N ₂) ^{**}	6357.63	790	5.6×10 ¹⁰	45	0.30
CD ₄ (N ₂ =1.5% H ₂ O) ^{**}	4991.26	50	1.4×10 ¹⁰	4.4	18
CH ₃ O (N ₂ =35% RB) ^{**}	2804.90	75	8.7×10 ¹⁰	7.2	0.12
CD (N ₂)	2190.66	50	3.3×10 ¹⁰	13	0.3
CD (prop) (low)	2190.66	50	7.4×10 ¹⁰	6.5	0.14
N ₂ O (air=9% RB) ^{**}	2193.63	50	1.5×10 ¹⁰	19	0.007
C ₂ H ₆ OH (N ₂) ^{**}	1934.2	770	2.2×10 ¹⁰	10	90
C ₂ H ₅ F ₂ (N ₂) ^{***}	1508.62	770	7.8×10 ¹⁰	0.6	0.009
NH ₃ (N ₂) ^{**}	1046.39	110	1.6×10 ¹⁰	20	0.006

Legend:
 ** Improved microstructure and double optical pass through ADSS
 *** With amplitude modulation and optical microstructure
 NNEA = normalized noise equivalent absorption coefficient
 NEC = noise equivalent concentration (σ available laser power and σ=1s) time constant, 18 dB vet filter slope

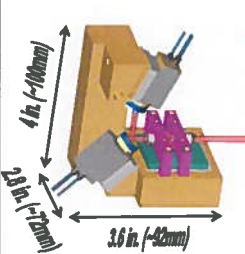
For comparison: conventional PAS 2.2 (2.4) × 10¹⁰ cm⁻¹W/Hz (1,800, 18,000 Hz) for NH₃ (σ=1s)

* M. B. Wallace et al. Appl. Opt. 47:2119-2126 (2008), ** J. S. Pappas et al. SAE Int. JCTN 2007-04-0112

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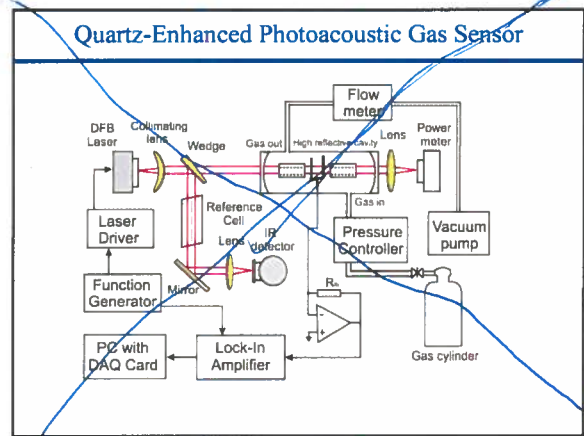
Future of Chemical Trace Gas Sensing

New design of fast broadly tunable EC-QCLs (2008)



- New optical configuration
Folded cavity (configuration #1)
- Fast tuning capabilities:
 - Coarse Broadband Scanning
($\sim 55 \text{ cm}^{-1}$ @ $5 \mu\text{m}$) **up to 5 KHz**
(compared to available technologies <math><10\text{Hz}</math>)
 - High resolution mode-hop free tuning ($\sim 3.2 \text{ cm}^{-1}$ @ $5 \mu\text{m}$) **up to 5 KHz**
(compared to available technology 100-200 Hz)

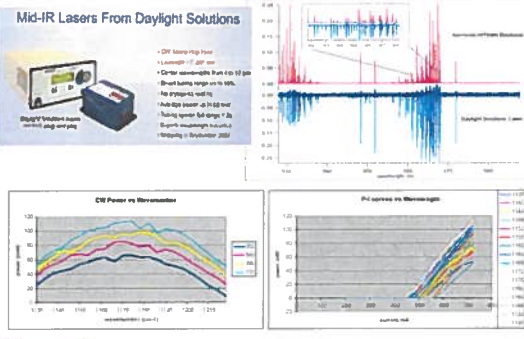
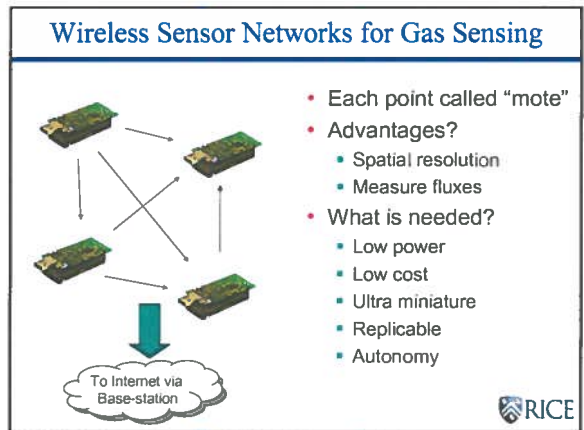
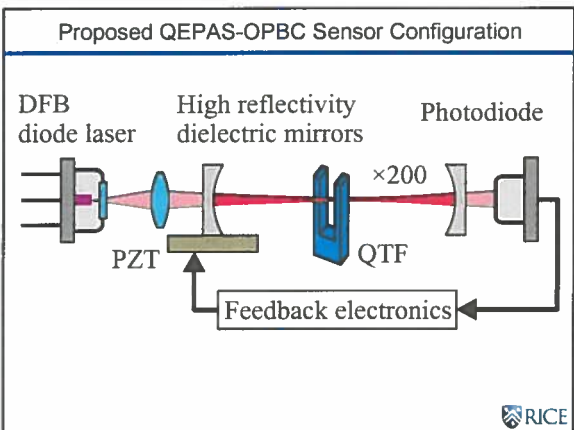
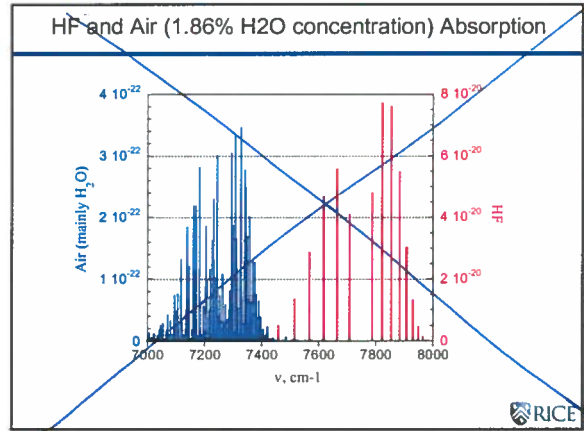
Patent pending, G. Wysocki, F. K. Tittel, 2007



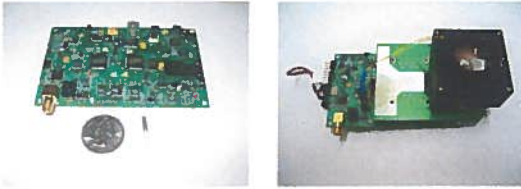
Widely tunable cw EC-QCL

Mid-IR Lasers From Daylight Solutions

- QW Masterchip laser
- Length: 7.5mm
- Output: approximately 100 mW @ 100 μm
- Broad tuning range: up to 100%
- No frequency lock-in
- Operates: up to 1000 hours
- Tuning speed: full range 7.5 μs
- Output wavelength: tunable
- Wavelength: 10.5 μm - 12.5 μm

Miniature QEPAS CO₂ sensor ($\lambda=2\mu\text{m}$) v2.0 boards

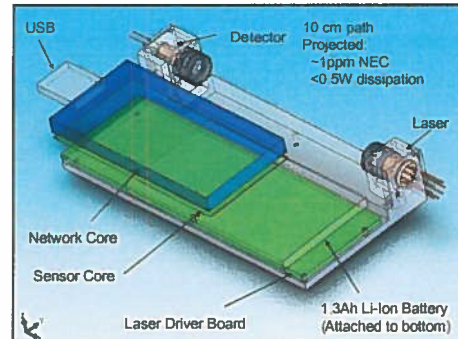


- Small size
- Relatively low cost
- High efficiency switching power supplies
- PWM Peltier cooler driver
- 0.2W control system power consumption
- Detection sensitivity* of CO₂ 110 ppm with 1sec. lock-in TC
- Over 10³ improvement in sensitivity is possible @ 4.2 μm

*G. Wysocki, A. A. Kostarev, and F. K. Tittel "Influence of Molecular Relaxation Dynamics on Quartz-Enhanced Photoacoustic Detection of CO₂ at $\lambda = 2\mu\text{m}$ ", *Applied Physics B* 85: 301-306 (2006)



Miniature LAS CO₂ sensor ($\lambda=2.7\mu\text{m}$) boards



Summary & Future Directions of QCL based Gas Sensor Technology

- **Quantum and Interband Cascade Laser based Trace Gas Sensors**
 - Compact, tunable, and robust
 - High sensitivity ($<10^{-4}$) and selectivity (3 to 500 MHz)
 - Capable of fast data acquisition and analysis
 - Detected 13 trace gases to date: NH₃, CH₄, N₂O, CO₂, CO, NO, H₂O, COS, C₂H₆, H₂CO, SO₂, C₂H₅OH, C₂HF₃ and several isotopic species of C, O, N and H.
- **New Applications of Trace Gas Detection**
 - Environmental Monitoring (urban quality - H₂CO and isotopic ratio measurements of CO₂ and CH₄, fire detection and quantification of engine exhausts)
 - Industrial process control and chemical analysis (NO, NH₃, H₂O, and H₂S)
 - Medical & biomedical diagnostics (NO, NH₃, N₂O, H₂CO and CH₃COCH₃)
 - Hand-held sensors and sensor network technologies (CO₂)
- **Future Directions and Collaborations**
 - Improvements of the existing sensing technologies using novel, thermoelectrically cooled, cw, high power, and broadly wavelength tunable mid-IR interband and intersubband quantum cascade lasers
 - New applications enabled by novel broadly wavelength tunable quantum cascade lasers based on heterogeneous EC-QCL (i.e. sensitive concentration measurements of broadband absorbers, in particular VOCs, HCs and multi-species detection)
 - Development of optically gas sensor networks based on QEPAS and LAS

