

Problem 10.9.8

KNOWN:

Diffusivity increases by a factor of two due to increase in temperature

FIND:

What must be the increase in temperature?

SCHEMATIC AND GIVEN DATA

STRATEGY:

Equations are available for diffusivities in liquid and gas phase as function of temperature. It is a straightforward application of those equations.

ASSUMPTIONS:

The same assumptions that go into the equations for diffusivities

SOLUTION:

1.) For Liquid $D = \frac{\kappa T}{6\pi\mu r}$ or we can say $D \propto T$.

$$\text{Therefore, } \frac{D_1}{D_2} = \frac{T_1}{T_2}.$$

To increase diffusivity by a factor of 2

$$\begin{aligned} \text{Final Temp. } (T_2) &= T_1 \times \frac{D_2}{D_1} = (273 + 10) \times 2 \\ &= 566K = 293^\circ C \end{aligned}$$

$$2.) \text{ For Gas } D_{AB} = \frac{0.001858 T^{3/2} (1/M_A + 1/M_B)^{1/2}}{p \sigma_{AB}^2 \Omega_{D,AB}}$$

or we can say $D \propto T^{3/2}$

$$\text{Therefore, } \frac{D_1}{D_2} = \frac{T_1^{3/2}}{T_2^{3/2}}$$

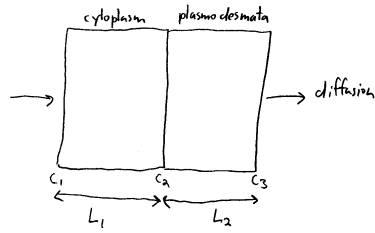
To increase diffusivity by a factor of 2,

$$\text{Final Temp. } (T_2) = T_1 \times \left(\frac{D_2}{D_1} \right)^{2/3} = 283 \times (2)^{2/3} = 449.23K = 176.23^\circ C$$

3.) Diffusion in gases is more easier than in liquids and increases more rapidly.

COMMENTS:

Problem 10.9.10



1)

Flux through cytoplasm = flux through plasmodesmata

$$j_{sugar, L_1+L_2} = -D_{sugar, cytoplasm} \times \frac{c_1 - c_2}{L_1}$$

$$= D_{sugar, plasmodesmata} \times \frac{c_2 - c_3}{L_2}$$

$$c_1 - c_2 = -j_{sugar, L_1+L_2} \frac{L_1}{D_{sugar, cytoplasm}} \quad \text{--- (1)}$$

$$c_2 - c_3 = -j_{sugar, L_1+L_2} \frac{L_2}{D_{sugar, plasmodesmata}} \quad \text{--- (2)}$$

Adding Eqns. 1 and 2,

$$c_1 - c_3 = -j_{sugar, L_1+L_2} \times \left[\frac{L_1}{D_{sugar, cytoplasm}} + \frac{L_2}{D_{sugar, plasmodesmata}} \right] \quad (3)$$

For an effective diffusivity giving the same flux due to the same concentration gradient

$$j_{sugar, L_1+L_2} = -D_{sugar, effective} \frac{dc}{dx}$$

$$= -D_{sugar, effective} \frac{c_1 - c_3}{L_1 + L_2}$$

$$\Rightarrow c_1 - c_3 = -j_{sugar, L_1+L_2} \times \left[\frac{L_1 + L_2}{D_{sugar, effective}} \right] \quad \text{--- (4)}$$

Equating (3) and (4)

Problem 10.9.10

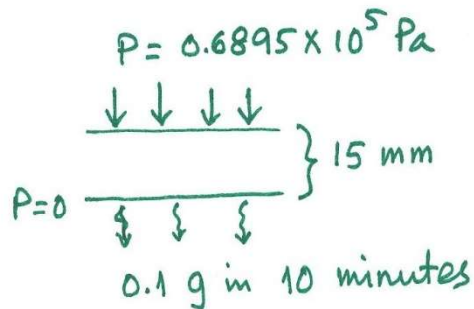
$$\begin{aligned}\frac{L_1 + L_2}{D_{sugar, effective}} &= \frac{L_1}{D_{sugar, cytoplasm}} + \frac{L_2}{D_{sugar, plasmodesmata}} \\ \frac{1}{D_{sugar, effective}} &= \frac{\frac{L_1}{L_1 + L_2}}{D_{sugar, cytoplasm}} + \frac{\frac{L_2}{L_1 + L_2}}{D_{sugar, plasmodesmata}} \\ &= \frac{f}{D_{sugar, cytoplasm}} + \frac{1-f}{D_{sugar, plasmodesmata}} \\ \frac{1}{D_{sugar, effective}} &= \frac{0.5}{5.6 \times 10^{-10} \text{ m}^2 / \text{s}} + \frac{1-0.5}{7.8 \times 10^{-12} \text{ m}^2 / \text{s}} \\ &= 6.5 \times 10^{10} \frac{\text{s}}{\text{m}^2} \\ D_{sugar, effective} &= 1.54 \times 10^{-11} \text{ m}^2 / \text{s}\end{aligned}$$

Problem 10. 9.12**KNOWN:**

Rate of flow and pressure gradient

FIND:

Intrinsic permeability

SCHEMATIC AND GIVEN DATA**STRATEGY:**

This is a straightforward application of Darcy's law that relates pressure gradient to the rate of flow.

ASSUMPTIONS:

1D flow; muscle dimension does not change as pressure is applied and water flows through it.

SOLUTION:

Using Darcy's Law, rewriting in terms of permeability

 $j^v =$ volumetric flux

$$\begin{aligned}
 &= -K \frac{\partial h}{\partial s} \\
 &= -\frac{k \rho g}{\mu} \frac{\partial h}{\partial s} \\
 &= -\frac{k}{\mu} \frac{\partial (\rho g h)}{\partial s} \\
 &= -\frac{k}{\mu} \frac{\partial p}{\partial s} \\
 &= -\frac{k}{\mu} \frac{\Delta p}{\Delta s}
 \end{aligned}$$

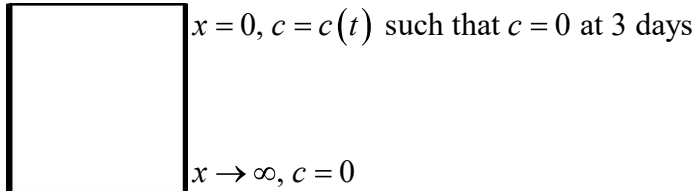
Q (volume collected) = Volumetric Flux x Area x time

$$\frac{0.1 \times 10^{-3} \text{ kg}}{10^3 \text{ kg/m}^3} = \frac{k \left[\text{m}^2 \right]}{0.001 \frac{\text{N} \cdot \text{s}}{\text{m}^2}} \frac{0.6895 \times 10^5 \text{ N/m}^2}{15 \times 10^{-3} \text{ m}} \cdot \frac{\pi}{4} (40 \times 10^{-3})^2 \text{ m}^2 (600 \text{ s})$$

$$k = 2.89 \times 10^{-17} \text{ m}^2$$

Problem 11.8.7

SCHEMATIC:



Governing equation:

$$\frac{\partial c}{\partial t} + u \frac{\partial c}{\partial x} = D \frac{\partial^2 c}{\partial x^2} - r$$

$u = 0$; no convection

$$\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2} - r$$

Boundary conditions:

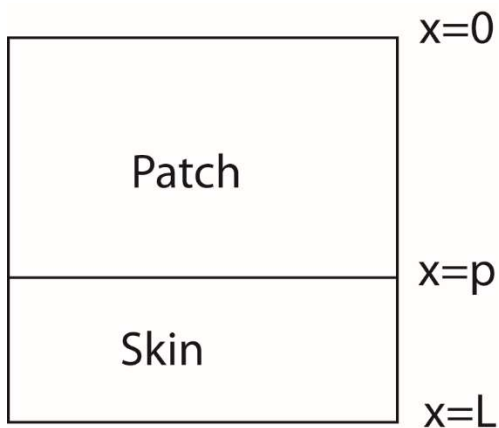
$$c(x=0) = c(t)$$

$$c(x \rightarrow \infty) = 0 \text{ or } -D \left. \frac{\partial c}{\partial x} \right|_{x \rightarrow \infty} = 0$$

Initial conditions:

$$c(t=0) = 0$$

Alternate formulation



Governing equation:

$$\frac{\partial c_p}{\partial t} = D \frac{\partial^2 c_p}{\partial x^2} \text{ In the patch}$$

$$\frac{\partial c_s}{\partial t} = D \frac{\partial^2 c_s}{\partial x^2} - r \text{ In the skin}$$

Boundary conditions:

$$-D_{\text{in patch}} \left. \frac{\partial c_p}{\partial x} \right|_{x=0} = 0$$

$$-D_{\text{in patch}} \left. \frac{\partial c_p}{\partial x} \right|_{x=p} = -D_{\text{in skin}} \left. \frac{\partial c_s}{\partial x} \right|_{x=p}$$

$$c_{p, x=p} = c_{s, x=p} \text{ assuming } K^* = 1$$

$$c_s(x=L) = 0$$

Initial conditions:

$$c_p(t=0) = c_i \text{ in the patch}$$

$$c_s(t=0) = 0 \text{ in the skin}$$

Problem 12.6.5**KNOWN:****FIND:****SCHEMATIC AND GIVEN DATA**

$$\text{G.E. } \frac{\partial c}{\partial t} + \mu \frac{\partial c}{\partial r} = D \frac{1}{r} \frac{d}{dr} \left(r \frac{dc}{dr} \right) - k''$$

$$\frac{1}{r} \frac{d}{dr} \left(r \frac{dc}{dr} \right) = \frac{k''}{D}$$

$$r = r_i \quad c = c_i \quad (1)$$

$$\text{B.C. } \begin{matrix} r = r_o & \frac{dc}{dr} = 0 \end{matrix} \quad (2)$$

STRATEGY:**ASSUMPTIONS:****SOLUTION:**

$$\frac{d}{dr} \left(r \frac{dc}{dr} \right) = \frac{k''}{D} r$$

$$r \frac{dc}{dr} = \frac{k''}{D} \frac{r^2}{2} + C_1$$

$$\text{Using (2): } 0 = \frac{k''}{D} \frac{r_o^2}{2} + C_1 \Rightarrow C_1 = -\frac{k''}{D} \frac{r_o^2}{2}$$

Integrating after rearranging

$$\frac{dc}{dr} = \frac{k''}{D} \frac{r}{2} - \frac{k''}{D} \frac{r_o^2}{2r}$$

$$c = \frac{k''}{2D} \frac{r^2}{2} - \frac{k''}{2D} r_o^2 \ln r + C_2$$

$$\text{Using (1): } C_2 = c_i - \frac{k''}{2D} \frac{r_i^2}{2} + \frac{k''}{2D} r_o^2 \ln r_i$$

$$c - c_i = \frac{k''}{4D} (r^2 - r_i^2) - \frac{k''}{2D} r_o^2 \ln r / r_i$$

Problem 12.6.5

At steady state, oxygen consumption is equal to oxygen inflow at the inner surface

$$\begin{aligned} &= -D \left. \frac{dc}{dr} \right|_{r=r_i} \\ \text{Flux} &= -\mathcal{D} \left[\frac{k''}{4\mathcal{D}} \cdot 2r_i - \frac{k''}{2\mathcal{D}} r_o^2 \cdot \frac{1}{r_i} \right] \\ &= -\frac{k''}{2} \left[r_i - \frac{r_o^2}{r_i} \right] \end{aligned}$$

$$\text{Flow} = \text{Flux} \times \text{Area} = 2\pi r_i L$$

$$\text{Flow/length} = \text{Flux} \times 2\pi r_i$$

$$\begin{aligned} &= -\frac{k''}{2} \left[r_i - \frac{r_o^2}{r_i} \right] 2\pi r_i \\ &= -k'' \pi (r_i^2 - r_o^2) \\ &= k'' \pi (r_o^2 - r_i^2) \left[\frac{\text{Flow}}{\text{m of vessel}} \right] \end{aligned}$$

COMMENTS: